2018

PHYSICS

(Major)

Paper: 5.3

Full Marks: 60

Time: 3 hours

The figures in the margin indicate full marks for the questions

GROUP-A

(Quantum Mechanics)

(Marks: 40)

- 1. Answer any four questions as directed: 1×4=4
 - (a) Select the correct answer:

 All the radiation laws can be shown to be special care of
 - (i) Wien's law
 - (ii) Rayleigh-Jeans law
 - (iii) Planck's law
 - (iv) Stefan-Boltzmann law

- (b) Which statement is correct?
 - (i) Phase velocity (v_p) of light wave is independent of λ in vacuum.
 - (ii) Phase velocity of matter depends on λ in vacuum.
 - (iii) Phase velocity of matter wave is independent on λ in vacuum.
 - (iv) Phase velocity of light wave is dependent on λ in vacuum.
- (c) What is the ground-state energy of a linear harmonic oscillator?
- (d) Show that

$$\left[x, \frac{\delta^2}{\delta x^2}\right] = -2\frac{\delta}{\delta x}$$

- (e) What is the total number of energy level (or degeneracy) for nth state of hydrogen atom?
- 2. Answer any three questions: 2×3=6
 - (a) A radio station operates at frequency of 103.7 Hz with a power output of 200 kW. Determine the rate of emission of quanta from the station.

- (b) What is the physical significance of the wave function $\psi(x, t)$?
- (c) Assume the uncertainty in the location of a particle is equal to its de Broglie wavelength. Show that the uncertainty in the velocity is equal to its velocity.
- (d) What is quantum mechanical tunnelling? Under what condition, the transmission coefficient T=1?
- (e) Draw the wave function of a particle in a box of infinite depth.
- 3. Answer any four questions:

5×4=20

(a) The energy distribution of blackbody radiation is given by Planck's law:

$$\rho(\lambda T) = 8\pi hc / \lambda^5 \frac{1}{\exp\left(\frac{hc}{\lambda kT}\right) - 1}$$

Show that for long wavelength

$$\rho(\lambda, T) \rightarrow 8\pi kT/\lambda^4$$

and for short wavelength

$$\rho(\lambda T) \to 8\pi hc / \lambda^5 \exp\left(\frac{-hc}{\lambda kT}\right)$$

What is Planck's quantum hypothesis?

Mention one experiment for determining

Planck's constant h. 3+1+1=5

(b) An α-particle is accelerated through a potential difference of 2000 volts. What is the wavelength of the associated de Broglie wave?

Given, mass of the proton =

 1.67×10^{-27} gms

Planck's constant

 $h = 6 \cdot 62 \times 10^{-27} \text{ erg sec}$

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1

- (c) (i) State and explain Heisenberg uncertainty principle.
 - (ii) Give an account of the γ ray microscope experiment.2+3=5
- (d) Explain the need for differential wave equation. Starting from the wave equation and introducing energy and momentum of the particle, obtain three-dimensional Schrödinger equation in time-dependent form. 2+3=5
- (e) (i) What is one-dimensional potential step?
 - (ii) A particle of mass m is moving in one-dimensional potential given by

$$V = \begin{cases} 0 & \text{for } x < 0 \\ v_0 & \text{for } x > 0 \end{cases}$$

If energy E of the incident particle is greater than v_0 , then calculate the coefficients of refraction and transmission. 2+2=4

(Continued)

4. Answer any two questions:

 $5 \times 2 = 10$

- (a) (i) What is an observable corresponding to a quantum mechanical system?
 - (ii) Establish the relation

$$[L_x, L_y] = i\hbar L_z; [L^2 L_z] = 0$$

where the notations have their usual meanings. What conclusion about the eigenfunction of the operators involved can be shown from those relation?

1+4=5

- (b) Discuss the wave mechanics of the electron in a hydrogen atom in a spherically symmetric potential and derive the energy state and energy function.

 2+3=5
- (c) Briefly discuss G. P. Thomson's experiment of electron diffraction, and its significance for quantum theory.

GROUP-B

(Astrophysics)

(Marks : 20)

- 5. Answer any three from the following: 2×3=6
 - (a) Draw a neat diagram of the celestial sphere showing a star in northern hemisphere, the celestial equator hour angle and the right ascension of the star.
 - (b) If one P-P chain transform 4.8×10^{-29} kg, then how many reaction cycles must produce the total transformed mass per second?
 - (c) What is universal time? Express 2165 sidereal days in terms of mean solar days.
 - (d) What do you mean by color index? What is the declination (δ) at celestial pole and celestial equator?
 1+1=2
 - (e) Calculate the temperature of Sun from the following data:

 $\lambda_m T = 0.287$, $\lambda_m = 4753$ Å

2

2

2

2

- **6.** Answer any two of the following: $4\times2=8$
 - (a) How are the spectra classified? What are the various spectral classes? Show that the colour of a star defines a spectral class. 1+1+2=4
 - (b) What is the main process that creates energy in solar system? Discuss P-P cycle. What is the end product of CNO cycle reaction under equilibrium condition? 1+2+1=4
 - (c) A star has a proper motion of 10 arc second per year. It is about 2 per sec away. The star radial velocity is measured to be 100 km/sec, i.e. it is moving towards the earth. Calculate star's space velocity.
- 7. Write short notes on any two of the following: $3\times2=6$
 - (a) Sidereal time
 - (b) Pulsars
 - (c) H-R diagram
 - (d) Black hole

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